

# Methyl 3-*para*-anisyl-4-(2-hydroxybenzoyl)isoxazole-5-carboxylate

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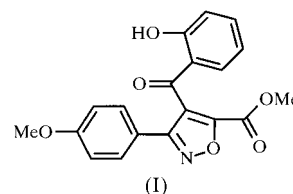
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The title compound, C<sub>19</sub>H<sub>15</sub>NO<sub>6</sub>, contains a planar isoxazole ring. An intramolecular hydrogen bond is formed between the OH group attached to a phenyl ring and a carbonyl O atom.

## Comment

In one of our research fields, the dipole–dipolarophile approach has been investigated. In the case of dipolar 1,3-cycloadditions, we have studied the reaction of diarylnitrilimines with dipolarophile derivatives of lindanone (Kerbal *et al.*, 1988), titralone (Theobald *et al.*, 1990) or lisochromanone (Kerbal *et al.*, 1990). Recently, our research has been focused on the reaction of aryl nitriloxides with isoquinoline arylidines (Badri *et al.*, 1999). We are trying to extend this research to the behaviour of new heterocyclic dipolarophiles, particularly that of esters of 4-oxo-4*H*-1-benzopyrancarboxylate with aryl nitriloxides. However, the isoxazolinic cycloadducts obtained so far seem to transform to more stable isoxazole products due to the opening of the pyronic cycle. Since NMR spectroscopy did not provide sufficient information about the regiochemistry of the reaction product we have carried out the X-ray structure analysis of the title compound, (I). The isoxazole ring of the title compound is planar (r.m.s. deviation = 0.004 Å) with the methoxycarbonyl moiety coplanar to it. The phenyl ring at C3 encloses an angle of 35.55 (3)° with the isoxazole ring. The other phenyl ring is nearly perpendicular [77.44 (5)°] to the isoxazole ring. An intramolecular hydrogen bond is formed between the hydroxyl group attached to a phenyl ring and a carbonyl O atom, so that a new six-membered ring is formed. Furthermore, there is a short intermolecular contact between the ester carbonyl O atom (O51) and the hydroxyl group.



## Experimental

4-Oxo-4*H*-1-benzopyran-2-carboxylate (9.8 mmol, 2 g) methyl ester and hydroxycarbonyl chloride (12 mmol) were mixed in anhydrous benzene (30 ml). The mixture was stirred, with regular addition of triethylamine (4 ml), for 15 min. After 2 h, the triethylamine chlorohydrate was eliminated and the solvent evaporated. The residue was then recrystallized from methanol.

### Crystal data

C<sub>19</sub>H<sub>15</sub>NO<sub>6</sub>  
M<sub>r</sub> = 353.32  
Monoclinic, *P*<sub>2</sub><sub>1</sub>/*n*  
*a* = 11.172 (1) Å  
*b* = 9.337 (1) Å  
*c* = 15.634 (1) Å  
*β* = 94.10 (1)°  
*V* = 1626.7 (2) Å<sup>3</sup>  
*Z* = 4

*D*<sub>x</sub> = 1.443 Mg m<sup>-3</sup>  
Mo *K*α radiation  
Cell parameters from 8192 reflections  
*θ* = 1–25°  
*μ* = 0.109 mm<sup>-1</sup>  
*T* = 173 (2) K  
Block, light yellow  
0.60 × 0.50 × 0.30 mm

### Data collection

Siemens CCD three-circle diffractometer  
*ω* scans  
Absorption correction: empirical (*SADABS*; Sheldrick, 1996)  
*T*<sub>min</sub> = 0.938, *T*<sub>max</sub> = 0.968  
32 334 measured reflections  
4240 independent reflections  
3444 reflections with *I* > 2σ(*I*)

*R*<sub>int</sub> = 0.030  
*θ*<sub>max</sub> = 29.57°  
*h* = –15 → 15  
*k* = –12 → 12  
*l* = –21 → 21  
400 standard reflections  
frequency: 960 min  
intensity decay: none

### Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.040  
*wR* (*F*<sup>2</sup>) = 0.105  
*S* = 1.024  
4240 reflections  
239 parameters  
H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0507P)^2 + 0.6770P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
(*Δ*/σ)<sub>max</sub> < 0.001  
*Δρ*<sub>max</sub> = 0.62 e Å<sup>-3</sup>  
*Δρ*<sub>min</sub> = –0.24 e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bonding geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
O43–H43...O51 <sup>i</sup>	0.91 (2)	2.61 (2)	3.0593 (14)	111.7 (16)
O43–H43...O41	0.91 (2)	1.80 (2)	2.6221 (13)	149 (2)

Symmetry code: (i)  $\frac{1}{2} - x, y - \frac{1}{2}, \frac{3}{2} - z$ .

All H atoms were initially located by difference Fourier synthesis. Subsequently, their positions were idealized and constrained to ride on their parent atoms with C–H(aromatic) = 0.95 Å or C–H(methyl) = 0.98 Å, and fixed individual displacement parameters [ $U(\text{H}) = 1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$  or  $1.2U_{\text{eq}}(\text{C})$ ]. The hydroxyl H atom was refined freely.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997).

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